

MATH327: Statistical Physics

Monday, 20 February 2023

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Something to consider

A micro-canonical system can't exchange energy with its surroundings.

How can we observe a system without exchanging energy with it?

Recap

Micro-canonical ensemble

Thermodynamic equilibrium

(dynamic)

Maximizing entropy

$$S = -\sum_i p_i \log p_i$$

(Second Law)

$$\rightarrow \log M$$

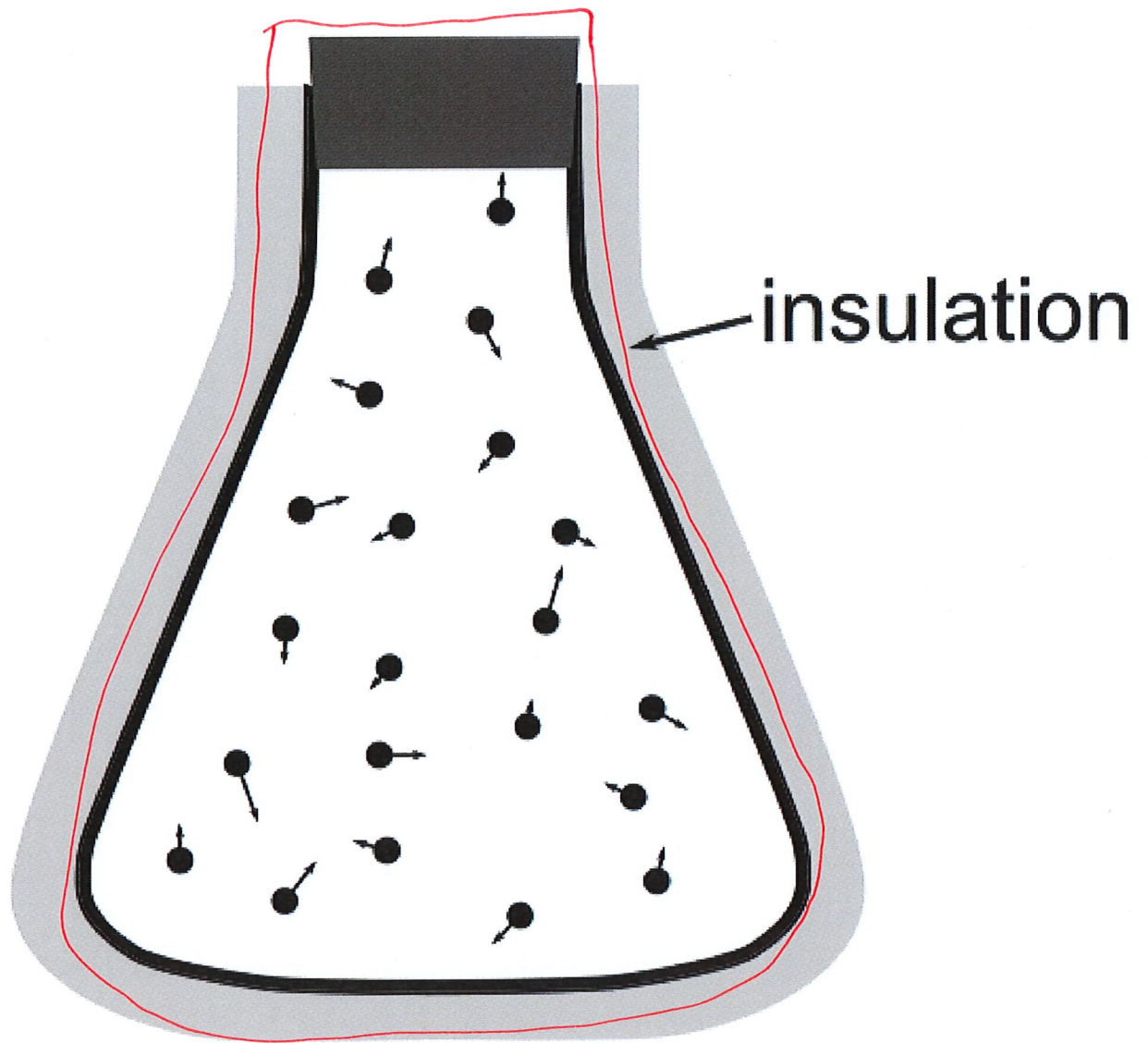
Plan

Temperature & heat exchange

Canonical ensemble

Partition function

Heat capacity



Microcanonical
(const. N E)

Temperature

Derived quantity should be stable in equil.

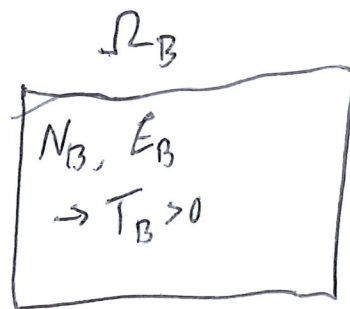
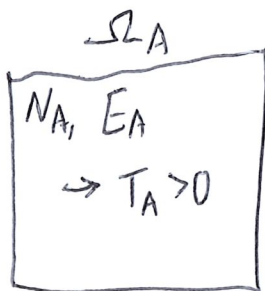
→ Function of conserved quantities $T(E, N)$

Define $\frac{1}{T} = \left. \frac{\partial S}{\partial E} \right|_N = \left. \frac{\partial}{\partial E} \log M \right|_N$

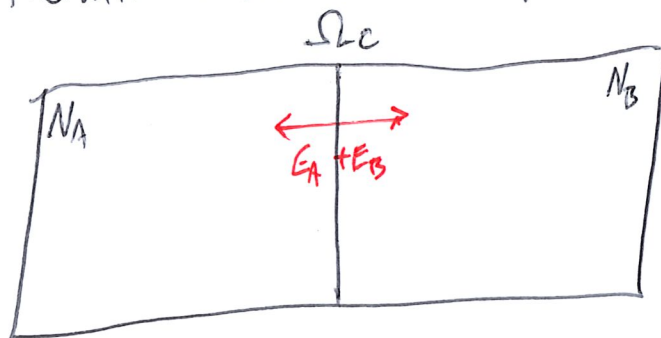
Adding energy → large increase in entropy
means small temperature

Heat exchange

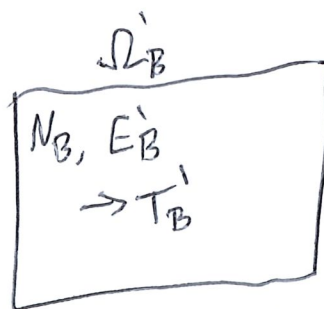
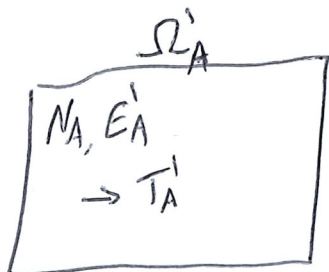
1) Two isolated micro-canonical systems



2) Put in thermal contact → equilibrate to $T_c > 0$



3) Re-isolate the two subsystems



Expect energy flow from hotter to colder system

Second law: $S(E_A) + S(E_B) \leq S(E_A + E_B) \leq S(E_A') + S(E_B')$

(1) (2) (3)

Let $E_s' = E_s + \Delta E_s$

$\Delta E_B = -\Delta E_A$

Taylor expansion $S(E_s') = S(E_s + \Delta E_s)$

$$= S(E_s) + \left. \frac{\partial S}{\partial E} \right|_{E_s} \Delta E_s + \mathcal{O}(\Delta E_s^2)$$

$$\approx S(E_s) + \frac{\Delta E_s}{T_s}$$

page 38

~~$S(E_A) + S(E_B) \leq S(E_A) + \frac{\Delta E_A}{T_A} + S(E_B) + \frac{\Delta E_B}{T_B}$~~

$$\frac{\Delta E_A}{T_A} - \frac{\Delta E_A}{T_B} \geq 0$$

$$\left(\frac{1}{T_A} - \frac{1}{T_B} \right) \Delta E_A \geq 0$$

page 39

$T_A > T_B \rightarrow \Delta E_A < 0$

Energy flows from hotter system Ω_A to colder system Ω_B

$T_A < T_B \rightarrow \Delta E_A > 0$

Same

" Ω_B
 Ω_A

$T_A = T_B \rightarrow \Delta E = 0$

Canonical ensemble

Complete isolation unrealistic

Instead consider canonical ensemble
characterized by fixed temperature T
and conserved particle # N

Fix temperature by thermal contact
with thermal reservoir ("heat bath")

System + reservoir remains micro-canonical

$$\Omega \otimes \Omega_{\text{res}} = \Omega_{\text{tot}}$$

$$E_{\text{tot}} = E + E_{\text{res}} \quad \text{conserved}$$

↑ fluctuate without changing
(intensive) T

Need to show details of Ω_{res} don't matter
→ consider Ω on its own

Replica trick

Ansatz: Let Ω_{res} be $R-1 \gg 1$ replicas of Ω
all in thermal contact & therm. equil.

$$E_{\text{tot}} = E + E_{\text{res}} = \sum_{r=1}^R E_r$$

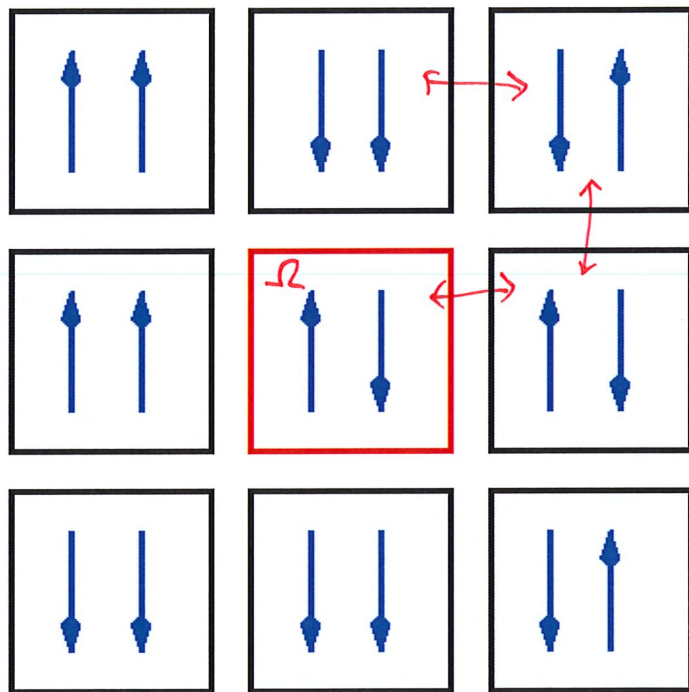
Let E_i be (non-conserved) energy of micro-state w_i of Ω

Occupation number n_i is # of replicas adopt w_i

$$\sum_{i=1}^M n_i = R$$

$$E_{\text{tot}} = \sum_{i=1}^M n_i E_i$$

$$\sum_i \frac{n_i}{R} = \sum_i p_i = 1 \quad \leftarrow \text{occupation probability}$$



w_i	n_i
$\uparrow\uparrow$	2
$\uparrow\downarrow$	2
$\downarrow\uparrow$	2
$\downarrow\downarrow$	3
$\sum_i n_i = 9 = R \checkmark$	
page 42	

System + reservoir Ω_{tot} fully specified by $M \{n_i\}$
or $\{P_i\}$

What is the temperature? $\frac{1}{T} = \left. \frac{\partial S_{tot}}{\partial E_{tot}} \right|_{N_{tot}} = \frac{\partial}{\partial E_{tot}} \log M_{tot} \Big|_N$

M_{tot} counts ways of arranging R replicas
into $\{n_i\}$ occupation #s

$$M_{tot} = \binom{R}{n_1} \binom{R-n_1}{n_2} \binom{R-n_1-n_2}{n_3} \dots$$

$$= \left(\frac{R!}{n_1! (R-n_1)!} \right) \left(\frac{(R-n_1)!}{n_2! (R-n_1-n_2)!} \right) \left(\frac{(R-n_1-n_2)!}{n_3! (R-n_1-n_2-n_3)!} \right) \dots$$

$$= \frac{R!}{n_1! n_2! n_3! \dots n_M!}$$

page 43

Entropy $S_{tot} = \log M_{tot} = \log(R!) - \sum_{i=1}^M \log(n_i!)$

Assume all $n_i \gg 1$ to apply Stirling's formula

$$S_{tot} \approx R \log R - R - \sum_i (n_i \log n_i - n_i)$$

$$= R \log R - R \sum_i P_i (\log P_i + \log R)$$

$$= -R \sum_i P_i \log P_i$$

$$n_i = P_i R$$

$$\sum P_i = 1$$

page 44

Thermodynamic equilibrium \leftrightarrow maximized entropy
conserving probability and E_{tot}

$$\bar{S} = -R \sum_i P_i \log P_i + \alpha \left(\sum_i P_i - 1 \right) - \beta \left(R \sum_i P_i E_i - E_{tot} \right)$$

$$\frac{\partial \bar{S}}{\partial P_k} = 0 = -R (\log P_k + 1) + \alpha - \beta R E_k$$

$$\log P_k = -1 + \frac{\alpha}{R} - \beta E_k$$

$$P_k = \exp\left[-\left(1 - \frac{x}{R}\right) - \beta E_k\right] = \frac{\exp(-\beta E_k)}{\exp\left(1 - \frac{x}{R}\right)}$$

$$= \frac{1}{Z} e^{-\beta E_k}$$

page 44

Impose constraints

$$1 = \sum_i P_i = \frac{1}{Z} \sum_i e^{-\beta E_i}$$

$$Z(\beta) = \sum_i e^{-\beta E_i}$$

partition function

$$E_{\text{tot}} = R \sum_i P_i E_i$$

$$S_{\text{tot}} = -R \sum_i P_i \log P_i = -R \sum_i P_i \log\left(\frac{1}{Z} e^{-\beta E_i}\right)$$

$$= R \log Z + R\beta \sum_i P_i E_i$$

$$= R \log Z + \beta E_{\text{tot}}$$

page 45

$$\frac{1}{T} = \frac{\partial S}{\partial E} = \beta + E \frac{\partial \beta}{\partial E} + \frac{R}{Z} \frac{\partial Z}{\partial \beta} \frac{\partial \beta}{\partial E}$$

$$\frac{1}{Z} \frac{\partial Z}{\partial \beta} = \frac{1}{Z} \sum_i \frac{\partial}{\partial \beta} e^{-\beta E_i} = -\frac{1}{Z} \sum_i E_i e^{-\beta E_i}$$

$$= -\sum_i P_i E_i = -\frac{E_{\text{tot}}}{R}$$

$$\frac{1}{T} = \beta + E \cancel{\frac{\partial \beta}{\partial E}} + R \left(\frac{-E}{R}\right) \cancel{\frac{\partial \beta}{\partial E}}$$

$$T = 1/\beta$$

We have derived the Gibbs distribution in therm. equil.

$$P_i = \frac{1}{Z} e^{-E_i/T}$$

$$Z = \sum_i e^{-E_i/T}$$

$$\frac{1}{T} = \beta$$

Boltzmann factor

Interpret p_i as probability system Ω adopt micro-state w_i with energy E_i (non-conserved)

Reservoir unknowable except for fixing temperature T

Derived quantities

Expectation value of internal energy

$$\langle E \rangle(T) = \sum_{i=1}^M E_i p_i = \frac{1}{Z} \sum_i E_i e^{-\beta E_i} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} = -\frac{\partial}{\partial \beta} \log Z$$

page 47

How does energy $\langle E \rangle$ depend on T ?

Heat capacity $C_V = \frac{\partial}{\partial T} \langle E \rangle \geq 0$

$(E_i - \langle E \rangle)^2$ Fluctuation-response relation

Higher temperature \rightarrow larger internal energy

$$\begin{aligned} \text{Entropy } S &= -\sum_i p_i \log p_i = -\sum_i p_i \log \left(\frac{1}{Z} e^{-\beta E_i} \right) \\ &= \log Z + \beta \sum_i p_i E_i = \log Z + \frac{\langle E \rangle}{T} \end{aligned}$$

page 48

$$\langle E \rangle = T \cdot S - T \log Z = T \cdot S + F$$

Helmholtz Free energy

$$F(T) = -T \log Z(T)$$

Rearranging,

$$Z = e^{-F/T}$$

$$p_i = \frac{1}{Z} e^{-E_i/T} = e^{(F-E_i)/T}$$

Derivatives of $F(T)$ related to $\langle E \rangle(T)$ and $S(T)$

$$\begin{aligned}\langle E \rangle &= -\frac{\partial}{\partial \beta} \log Z = \frac{\partial}{\partial \beta} \left(\frac{F}{T} \right) = \frac{\partial T}{\partial \beta} \frac{\partial}{\partial T} \left(\frac{F}{T} \right) \\ &= \frac{-1}{\beta^2} \frac{\partial}{\partial T} \left(\frac{F}{T} \right) = -T^2 \frac{\partial}{\partial T} \left(\frac{F}{T} \right)\end{aligned}$$

Similarly $\frac{\partial}{\partial T} F = \frac{\partial}{\partial T} (-T \log Z) = -\log Z - T \frac{\partial}{\partial T} \log Z$

$$= -\log Z - \frac{\langle E \rangle}{T} = -S$$

$$\begin{aligned}S &= -\frac{\partial}{\partial T} F \\ &= \frac{\langle E \rangle - F}{T}\end{aligned}$$