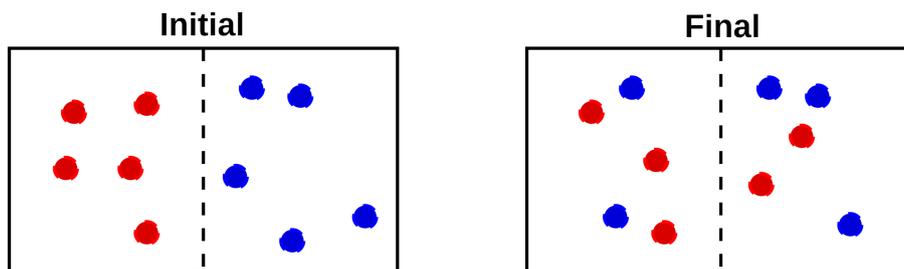


# MATH327: Statistical Physics, Spring 2022

## Tutorial problem — Mixing entropy

Let's consider a slight variation to the particle exchange thought experiment we worked through in class. We again begin with two canonical ideal gases, initially separated by a wall, each with  $N$  particles in volume  $V$  at temperature  $T$ . **All**  $2N$  particles have identical physical properties, *except* that those initially in the left compartment (the “reds”) are distinguishable from those in right compartment (the “blues”) by their colour. Call this initial system  $\Omega_0$ . We have already computed its entropy  $S_0 = 2S_I(N, V) = 5N + 2N \log \left( \frac{V}{N\lambda_{\text{th}}^3} \right)$ , where  $\lambda_{\text{th}} = \sqrt{2\pi\hbar^2/(mT)}$ .

We then carry out the procedure of removing the wall, allowing the combined system to reach thermodynamic equilibrium, and then re-inserting the wall to re-separate the two systems. Call the combined system  $\Omega_C$  with entropy  $S_C$ . As discussed in class, it's safe to assume that  $N$  particles end up in each of the two re-separated systems. However, red and blue particles can now appear in either of the two re-separated systems. Call this final system  $\Omega_F$  with entropy  $S_F$ . The initial and final systems are illustrated by the figure below.



The first task is to compute the mixing entropy  $S_{\text{mix}} = S_C - S_0$ , where in the combined system  $\Omega_C$  we now have two sets of  $N$  indistinguishable particles, but can distinguish between the two sets. The starting point is the partition function

$$Z_C = \frac{1}{N!} \frac{1}{N!} Z_1^{2N} = \frac{1}{N!} \frac{1}{N!} \left( \frac{2V}{\lambda_{\text{th}}^3} \right)^{2N},$$

where  $Z_1 = 2V/\lambda_{\text{th}}^3$  is the single-particle partition function.

The second task is to compute the final entropy  $S_F$ , to see whether  $S_F \geq S_C$  as demanded by the second law of thermodynamics. We can break this up into two steps. The first of these is to compute the partition function  $Z_F$  of the two re-separated systems (each with  $N$  particles), summing over all ways of dividing the red and blue particles between them. The following special case of the [Zhu–Vandermonde identity](#) may be useful for this step:

$$\sum_{k=0}^N \binom{N}{k}^2 = \binom{2N}{N}.$$

Finally, use your result for  $Z_F$  to determine the final entropy  $S_F$ .